OS92 2002 Ocean Sciences Meeting

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United States The non-linear responses of marine optical signals have made coastal ocean areas of Case 2-type waters a challenging environment for remote sensing. Hy-perspectral remote sensing with its continuous, high-resolution spectral information has long promised to help in unraveling some of the difficulties by bring-ing to bear the mathematical tools of imaging spec-troscopy onto the non-linear problem. However, these tools require a high confidence in the absolute radio-metric calibration of the hyperspectral sensor. Dur-ing the 2001 Hyperspectral Coastal Ocean Dynamics Experiment (HyCODE) at the Long-term Ecological Observatory-15 m (LEO-15) site off the coast of New Jersey, we collected multiple days of high altitude im-agery in support of the ONR objectives to develop in-situ optical hyperspectral algorithms and nowcast-forecast techniques. An explanation of the calibration techniques and data produced by the Portable Hyper-spectral Imager for Low Light Spectroscopy II (PHILLS II) will be presented, as well as comparisons between II) will be presented, as well as comparisons betw the hyperspectral imagery and in-situ data

OS21A-14 0830h POSTER

- Application of Remote Sensing Multitemporal/Multisensor Data Analysis and GIS Database for Coastline Change Monitoring and Nearshore Morphology Detection in Rio Grande do Norte State, Northeast Brazil
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The objective of this study was to define an operational methodology using remote sensing and geographic information system techniques for monitoring and predicting shoreline change and nearshore composition/structure identification in northeast Brazilian EW-oriented coastline. This area is inserted on the PETROBRAS oil exploration research. The multitemporal approach used remote sensing technology either on a large scale with the most aerial photos and medium scale with SPOT-HRV/HRVIR and Landsat TM/ETM+ satellite data integrated through a GIS database with ancillary maps (eg. topography, bathymetry, geology), physical parameters (eg. curvents and wind velocity/direction, tidal observation, beach profile) and sediment analysis (eg. in situ vanven collected samples). The temporal images comparison method (vi-The objective of this study was to define an op-

veen collected samples). The temporal images comparison method (vi-sual/digital) was applied for qualitative and semi-quantitative (images spatial resolution dependent) data analysis of sediment budget to this coastal system. The results showed that most accretion areas are due to sed-iment capture on E-W oriented sand spits while erosion areas are linked to large scale bottom morphology. The changes are mainly due to longshore drift contributions and negrine sediment budget to the first sediment budget to the sediment budget to the sediment budget of the sediment and negative sediment budget.

and negative sediment budget. Knowing that detecting features on remotely sensed imagery is dependent upon the type of targets on sur-face, size, association and tone distribution, many digi-tal image processing procedures were tested to provide enhanced images to a properly interpretation. Using RGB and IHS color composites allowed to distinguish between carbonate, terrigenous and mixed sands de-posits in nearshore until around 25 meters depth. Sea-water in this coastal region was very clear at the time of

satellite scanning path. Highly turbid water was con-centrated in a zone a few hundred meters large beside beach line through which huge quantities of sand and clay are transported to the west by currents parallel the shoreline. The sediment character distribution were confirmed by analysis of vanveen collected samples. High-pass filtering applied to single visible or infrared bands highlighted some important offshore morphologic features that can represent successive beachrocks lines, sandbars strongly oriented probably sustained by struc-turally controlled blocks edges, as well as sand waves highly dependent upon the deep currents. Regional lin-eaments maps of coastal zone obtained by directional filtered images combined with actual land forms fea-tures and drainage system showed a indisputable evisatellite scanning path. Highly turbid water was con tures and drainage system showed a indisputable evidence of continuity offshore of these morphostructural lineaments

lineaments. The study confirmed that remote sensing and GIS integration techniques are essential tools for shoreline morphodynamic controls, nearshore composition detec-tion, offshore features identification, monitoring and predicting onshore-offshore sediment budget balance on cvclic movement.

OS21B HC: Hall III Tuesday 0830h

Novel Techniques for Chemical **Characterization in Marine Systems** II

Presiding: H E Hartnett, Rutgers University; L Minor, Old Dominion University

OS21B-15 0830h POSTER

Mass spectrometric characterization of 13C-tracers: applications for biogeochemical study

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University of Georgia, Department of Marine Sci-ences, UGA, Athens, GA 30602 Stable isotope tracers have been widely used to study biogeochemical cycles of carbon and nutrients. A new mass spectrometric approach was developed in my lab to characterize 13C-lipids in tracer microcosm ex-periments. The principal of this approach was based on the quantitative shifts in m/z of 13C-lipids characteris-tic fragments. The shifts vary with carbon number and labeling content in these fragments. In biogeochemi-cal tracer experiments, the 13C-labeled (uniformly or specifically) lipids can be readily distinguished from natural counterparts by calibrating with a series of mixtures of 13C-labeled and unlabeled standards. New 13C-labeled lipid compounds produced from organic matter degradation can be monitored by examining 13C fragments in their mass spectra. An advantage of this approach is the capability of examining intra-molecular relationship between organic substrates and metabolism products by analyzing details of mass spe-tra. For example, analysis for labeling content of newly produced compounds (partially- or uniformly-labeled) indicated two different metabolism pathways: resyn-thesis and inter-transform from substrates. This ap-proach has been successfully applied to study (1) effects of redox conditions on organic matter degradation, (2) role of benthic macrofauna in sediment diagenesis of al-gal material, and (3) influences of biological and phys-ical mixing processes on organic carbon cycling.

OS21B-16 0830h POSTER

Nanomolar Detection for Phosphate and Nitrate Using Liquid Waveguide Technology.

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Hoe, Plymouth PL1 3DH, United Kingdom The ability to detect ambient concentrations of ni-trate and phosphate is of vital importance in under-standing the cycling of these nutrients, particularly in the oligotrophic regions of the worlds oceans. Various methods and novel technologies have been employed over recent years to address the problems of analysing these nutrients at the ambient nanomolar concentra-tions found in oligotrophic oceanic regions. With ad-vances in long path-length Liquid Waveguide Capillary Cells there is the ability to use these in conjunction with sensitive segmented flow colorimetric analysis sys-tems to produce analytical methods for the nanomolar detection of nitrate, nitrite and phosphate. Prelimi-nary data are presented here for phosphate concentra-tions from samples analysed during a cruise to the nu-trient deplete Eastern Mediterranean Cyprus Gyre re-gion, with a detection limit for phosphate of less than 2 nanomoles per litre. Also reported are nitrate results

from the surface waters of the oligotrophic Northern Indian Ocean with a detection limit of 1 nanomole p

OS21B-17 0830h POSTER

Iron Isotopic Compostion of Marine Samples

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Edward A Boyle¹ ((617)253-3388; eaboyle@mit.edu)

¹MIT, Dept of Earth, Atmospheric, and Planetary Sciences, 77 Mass Ave, Rm E34-266, Cambridge, MA 02139, United States Iron is an essential micronutrient in the ocean and

MA 02139, United States Iron is an essential micronutrient in the ocean and a limiting nutrient in high nitrate, low chlorophyll (HNLC) regions of the ocean. Although the importance of iron in the ocean has been recognized in the past decade, it is difficult to study because of its complex chemistry and behavior, and the difficulty in obtaining measurements without contamination. Fractionation of iron isotopes could be an effective tool to investigate and quantify the marine geochemistry of iron. Ini-tial studies of iron isotopes show measurable fraction-ation in both abiotic and biological processes (Bullen and McMahon, 1997; Beard and Johnson, 1999; Zhu et al., 2000; Anbar et al., 2000; Belshaw et al., 2000). For example, a 1.4 permil (56/54 Fe) variation in iron isotopes of seawater over the past 7 Ma has been in-ferred from a paleorecord of iron isotopes reconstructed from a Fe-Mn nodule (Zhu et al., 2000). This study will address questions about the iron cycle using di-rect measurements of stable Fe isotopes in marine sam-ples. Trace metal clean plankton tows were collected in the tropical eastern Atlantic (10° N, 45.5° W), filtered, and then measured for their iron isotopic composition. Measurements were made using a Micromass IsoProbe Multi-collector ICPMS. This system uses a hexapole collision cell to reduce molecular interferences and im-prove transmission. Initial results using a microflow Multi-collector ICPMS. This system uses a hexapole collision cell to reduce molecular interferences and im-prove transmission. Initial results using a microflow PFA nebulizer, argon and hydrogen collision gases, and standard-sample bracketing give an external precision better than 0.2 (2 σ) permil in the 56/54 Fe. Iron was mobilized from plankton tow samples by two different digestion methods: nitric acid/hydrogen peroxide di-gestion and muffle furnace combustion. Then the iron was purified by anion exchange chromatography. Repli-cates of one plankton tow sample have an iron isotopic value that is 0.42 \pm 0.2 (2 σ) permil depleted relative to measured igneous rock samples (56/54 Fe ratio). This measurement demonstrates that iron in the upper ocean is fractionated from continental rock sources. It is pos-sible that this iron isotope fractionation is due to ma-rine organisms, although other possibilities must first be ruled out (e.g. fractionation during release from dust). If the iron fractionation is biological, iron ex-port from the euphotic zone should lead to progressive light isotope depletion, similar to the carbon and nitro-gen isotope systems. Iron stable isotope measurements may thus yield information about the degree of iron utilization in the upper ocean.

OS21B-18 0830h POSTER

The Use of Staining Techniques and Flow Cytometry to Identify and Isolate POM Subclasses for Organic Mass Spectrometric Analyses

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Claire Faust¹ (757-683-6080)

¹Department of Chemistry and Biochemistry, Old Do-

¹ Department of Chemistry and Biochemistry, Old Dominion University, Norfolk, VA 23529 Flow cytometric sorting is a useful technique for identifying and isolating sub-populations of particles within natural particulate organic matter (POM) samples. It has recently been used to isolate phytoplankton and non-phytoplankton "detrital" particles on the basis of chlorophyll autofluorescence and forward light scatter. Because autofluorescence and sevenation or staining techniques were necessary. Therefore, by applying direct temperature-resolved mass spectrometry (DT-MS) to the sorted subclasses, a comparison of the chemical characteristics of "phytoplankton" and "detritus" was made(1,2). While this is a useful initial approach for demonstrating the morphological and chemical heterogeneity of water-column POM, it suffers from distinct limitations. Non-fluorescent cells, such as heterotrophic bacteria and zooplankton (that have not just eaten a phytoplankton cell) are included within the "detritus" was the food.

heterotrophic bacteria and zooplankton (that have not just eaten a phytoplankton cell) are included within the "detrital" pool of POM. In this study we attempt to include these cells within our POM subclasses. In order to identify bacterial and eukaryotic cells within natural aquatic particle samples, various nucleic acid stains (SYTO 16, DAPI, SYTOX, etc.) are be-ing tested. In addition to optimizing staining times and concentrations, we are examining the effects of the staining procedures on the molecular-level characteris-tics of the particles. Although the relative sensitivi-ties of the stains to various cell and virus types need

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to be further constrained, initial results look promisto be further constrained, initial results look promis-ing. Pyrolysis-gas chromatography-mass spectrometry shows that the membrane permeable nucleic acid stain SYTO 16 does not alter the molecular-level characteris-tics of POM. Therefore it appears possible to use stain-ing and flow cytometry to identify the cells, flow cy-tometric sorting to isolate them from the rest of the natural aquatic particles, and DT-MS to determine the chemical characteristics of the cell vs. non-cell particle populations. populatio

(1) Minor, E.C., T.I. Eglinton, R. Olson, J.J. Boon (1998). Organic Geochemistry 29(5-7): 1561-1582.
(2) Minor, E.C., T.I. Eglinton, J.J. Boon, and R. Olson (1999). Analytical Chemistry 71(10): 2003-001101 2013

OS21B-19 0830h POSTER

A New Rapid Chemical Method for Analyzing Si, Ca, Al, and Other Elements in Marine/Freshwater Particles and Sediments.

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smanganini@whoi.edu); Lary A. Ball¹; Kazuhiro Hayashi²; James H. Churchill¹; <u>Rachel H. Stanley¹</u>; Susumu Honjo¹

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A new chemical method has been developed to de-A new chemical method has been developed to de-termine Si, Ca, Al, and other elements (Fe, Mg, Sr, Mn, Ba, P, V, Ni, Cu, Ti, S) in particles and sedi-ments from marine and freshwater environments. The procedure has been tested on sediment trap materials, sediments and suspended particles that have been col-lected on polycarbonate and polyethersulphone filters sectiments and suspended particles that have been col-lected on polycarbonate and polyethersulphone filters with the use of in-situ filtering equipment. The pro-cedure uses Teflon pressure vessels and microwave oven heating to dissolve the samples in a HNO3/HF solution. Variable acid and pressure conditions are explored to insure complete sample dissolution and to prevent Ca precipitation and Si volatilization in the presents of 10mg sample in 15 minutes. The solutions are prepared for the ICP-MS with the addition of I as an internal standard and for the ICP-ES with the addition of scan-dium Sc as the internal standard. A Teflon/sapphire nebulizing unit is used to quantify Si in the presents of HF. The respective precision (cv) and accuracy as re-covery rates (percent) of the following elements in ref-erence solutions are presented: (Si, 4.61, 103.1), (Ca, 3.63, 99.8), (Al, 6.39, 101.5), (Ba, 1.64, 95.7), (Fe, 7.37, 104.2), (Mg, 8.40, 100.1), (Sr, 5.74, 99.8).

OS21B-20 0830h POSTER

Sensitive Determination of Iodide Species in Fresh or Saline Matrices Using High Performance Liquid Chromotography and UV/Visible Detection

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U, Galveston, TX 77551. United States Iodide is determined directly using a new method combining anion exchange chromatography and ultravi-olet visibility detection. Iodate and the total of organic iodine species are determined as iodide with minimal sample preparation. The method has been successfully applied to mea-sure the concentrations of iodide species in rains, fresh surface waters and ground waters, estuarine waters of different salinities, and seawaters. The detection limit is 0.4 ppb with less than 5% relative standard devia-tion. Analytical accuracy was tested against certified reference material SRM 1549, powdered milk (NIST) and through the method of standard additions. The organic decomposition technique yielded recoveries of L-thyroxine, SRM 1549 milk, and iodinated humic and fulvic materials.

OS21B-21 0830h POSTER

Application of X-ray photoelectron spectroscopy (XPS) and density fractionation to study organic-mineral interactions in marine sediments

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Interactions between organic matter and minerals Interactions between organic matter and minerals are important for the preservation of organic mate-rials in marine sediments. Recent evidence suggests that these interactions may not be due to protection of the organic matter via simple sorption to miner-als, but rather that the organic matter is acting as a glue between mineral particles resulting in aggregate formation. Density fractionation into multiple density intervals is a powerful approach for investigating the patchy nature of organic matter distribution because isolates with varying oreanic matter loadings can be patchy nature of organic matter distribution because isolates with varying organic matter loadings can be obtained. The aggregate surfaces can be examined us-ing X-ray photoelectron spectroscopy (XPS), since this technique is restricted to the top 10 nm of the sample surface. XPS gives quantitative information on all the elements on the surface, except H and He, and in addi-tion, gives some information on the chemical bonding of carbon and nitrogen. We used these techniques, along with more standard geochemical tools to study sedicarbon and nitrogen. We used these techniques, along with more standard geochemical tools, to study sedi-ment samples along two transects: one off the coast of Washington, and one off the western coast of Mexico. These transects span a wide range of oxygen exposure times (OET), from a few days in the oxygen deficient zone off Mexico, to a thousand years in the deepest sites off Washington. We will present our results and describe how XPS can be used to study changes in ox-idation state of the organic matter along the transects and how that relates to OET. Furthermore, the poten-tial of XPS to give information on the location of the OM on the mineral surfaces will be discussed.

OS21B-22 0830h POSTER

A high temperature combustion instrument comparison for measurement of dissolved organic nitrogen in seawater

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In a recent comparison of total dissolved nitro-gen (TDN) in seawater, relatively good agreement was found among 30 analyses using a variety of UV oxida-tion, persulfate oxidation and high temperature comtion, persulfate oxidation and high temperature com-bustion (HTC) methods; but the resultant dissolved or ganic nitrogen (DON) agreement was not good enough. Since the tightest agreement and best prospects for easy future analyses lay with HTC, further evaluation of these methods was pursued. We assembled instru-ments in one location to investigate similarities and differences in combustion systems, plumbing, and ni-trogen detection. Upon establishing comparability in recovery of inorganic and organic nitrogen compounds and deciding upon modifications of individual instru-ments, authors returned their home laboratories. A se-ries of samples from estuarine to deep ocean were col-lected that varied in TDN from about 5 to 80 uM N, with 20 to 95 percent of the TDN representing DON. This comparison used 4 stand-alone instruments (man-ufactured by Antek, Dohrmann, Shimadzu, Skalar) and This comparison used 4 stand-alone instruments (man-ufactured by Antek, Dohrmann, Shimadzu, Skalar) and 4 hybrid setups with combustion through Shimadzu TOC analyzers linked to Antek, Seivers, Skalar, and Yanaco NO analyzers. Results will be presented of TDN, DON, and DOC from this comparison.

OS21B-23 0830h POSTER

Development of Compact UV Laser Induced Fluorescence Instrument and Preliminary Measurements of Organics in the Gulf of Mexico

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phy 114, Tampa, Fl 33620, United States We have developed a high-PRF, 266 nm laser based Laser Induced Fluorescence (LIF) instrument that is 1000 times more sensitive than existing in situ fluo-rescence sensors (SAFire) for the detection of colored dissolved organic matter (CDOM) in ocean water. The increased sensitivity is due to the use of a pulsed UV laser instead of a continuous spectral lamp for excita-tion, and the ability to use high-PRF boxcar integra-tors and sensitive optical photomultiplier tubes to de-tect and average the weak CDOM fluorescence signal. Our portable LIF system consisted of a (4x) Nd:YAG microchip laser source (5 mW, 8 kHz) operating at 266 nm, 13 optical filter channels ranging from 239 nm to 685 nm in combination with absorption filters with cut-

microchip laser source (5 mW, 8 kHz) operating at 266 nm, 13 optical filter channels ranging from 239 nm to 685 nm in combination with absorption filters with cut-off at 280 nm and 295 nm, and a PMT detector and a boxcar integrator averager system. A laptop computer was used for data acquisition (via GPIB interface) and controlling the filter wheels. A software program was written in Labview to acquire, store and monitor the optical channels in real-time. The emission channels were switched every few seconds, resulting in a complete scan in about one minute. The transition the Gulf of Mexico. The instrument sensitivity was tested in a variety of water masses using the ships underway water supply. The fluorescence signal from CDOM near 450 nm originating from clean seawater 100 miles off-shore in the Gulf of Mexico had a signal-to-noise of greater than 120. To compare the detection limits of the system, quinine sulfate was used as the fluorescence standard. The detection limit of the portable LIF system was measured to be 0.001 ppb of quinine sulfate. It was determined that the sensitivity SPEX; Lab system) was 0.05 ppb of quinine sulfate.

OS21B-24 0830h POSTER

Preparation of Reference Material for Nutrients in Seawater (RMNS) on the Seawater Base

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"Meteorological Research Institute, Nagamine 1-1,Tsukuba, IBARAKI 305-052, Japan Kansai Environmental Engineering Center Co., Ltd. has carried out the oceanographic survey in the western North Pacific Ocean since 1990. We are carrying out the long-term monitoring observation with high qual-ity at research cruises in the open ocean. Nutrient mea-surements are one of the most important components of nitrate measurement during the WOCE, the required accuracy was better than 1percent, and the repeatabil-ity (precision) was 0.2percent. In order to improve the quality of nutrients measurements, it is necessary to make reference materials with high quality. Reference materials for nutrients in seawater (RMNS) on the sea-water base were developed by using autoclave. The sea-water base unter developed by using autoclave. The sea-water base autoclaved in a huge autoclave (W,145cm H,155cm D,290cm) at 120deg., 2 hours, 2 times. Polypropylene bottles of 100 ml capacity and the polypropylene caps were also autoclaved before filling the seawater. The autoclaved seawater of 90 ml was filled in polypropy-lene bottles, and. Ppolypropylene bottle was sealed in a clean bench in a clean room. Since the stainless stel container entered into the huge autoclave from the nor-mal environment side and taken out to the clean envi-ronment side after the sterilization, the degree of clean-ness of the autoclaved seawater was ensured. The ho-mogeneity of our RMNS is as follows and the standard deviations of the analytical results were same as the repeatability of the measurements of the same sample. Example of analysis of RMNS (microM at 20deg.) Bottle No NO3 NO2 SiO2 PO4 No 1 17.47 0.95 153.8 1.43 No 2 17.56 0.96 154.0 1.4 Kansai Environmental Engineering Center Co., Ltd.

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deliver our RMNS worldwide, we could make a pier to start new ages when nutrients analysis is easily carried out with better traceability and high quality. We are going to start a cooperative measurement in some labo-ratories on land and multiple research cruises using our DMNSC RMNS.

OS21B-25 0830h POSTER

Reference Material for Nutrients in Seawater: Results of Long-term Storage Experiments and how to and who Certify it?

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We have been carried out the long-term storage ex-We have been carried out the long-term storage ex-periments of the reference material for nutrients in sea-water (RMNS) for up to 4 years. The importance of reference material for nutrients in seawater is now well recognized. The needs of RMNS are increasing in in-ternational hydrographic program to have better trace-ability as well as a basis for quality control and quality assurance in the measurements of nutrients in seawatermational hydrographic program to have better trace-ability as well as a basis for quality control and quality assurance in the measurements of nutrients in seawa-ter. Our approach to preparing RMNS using autoclav-ing was developed and tested by previous researchers. We believe that to supply our RMNS to the oceano-graphic community worldwide, the RMNS should be stable more than 2-3 years because of the reasons as fol-lows. 1) The several months leading time before ship-ping is needed to test the homogeneity of the RMNS. 2) To establish the traceability between batches of RMNS, it will also take some months. 3) During the several years period of international program, it is better to use less number of batches. 4) Although, we can not certify the RMNS yet at the present time, it will take several months to put the concentrations of the RMNS and is-sue the certificate of the RMNS. The details on the preparation of our RMNS will be presented separately by one of the co-authors of this study in this meet-ing. The samples for long-term storage experiment are prepared in November 1996 and the latest measurement was done in Feb. 2001. Two independent measurements were done just after preparation to put the initial val-ues for nitrate, nitrite, phosphate and silicate. Kansai Environmental Engineering Center (KANSO) put 23.07 H/- 0.13 umol/l for nitrate, 1.27 +/- 0.01 umol/l for nitrite, 0.42 +/- 0.07 umol/l for phosphate and 57.09 +/- 0.34 umol/l for silicate respectively. Japan me-teorological Agency (JMA) put 23.41 +/- 0.04 umol/l for nitrate, 1.28 +/- 0.01 umol/l for nitrite, 0.41 +/-0.03 umol/l for phosphate and 57.57 +/- 0.99 umol/l for silicate, respectively. Two independent measure-ments were in good agreement, then we could confirm the homogeneity of our RMNS and good traceability between two laboratories. The results of our long-term the homogeneity of our RMNS and good traceability between two laboratories. The results of our long-term experiments showed that homogeneity and stability are maintained for about 4 years. JMA put 22.86 +/- 0.14 umol/l for nitrate, 1.30 +/- 0.01 umol/l for nitrite and 0.39 +/- 0.03 umol/l for phosphate in July 2000, re-spectively. KANSO put 23.33 +/- 0.42 umol/l for ni-trate, 1.27 +/- 0.01 umol/l for nitrite, 0.39 +/- 0.03 umol/l for phosphate and 56.92 +/- 0.73 umol/l for sil-icate in Feb. 2001, respectively. Our results, then, had demonstrated that the RMNS can be obtained using autoclaving and can be stored for about up to 4 years. We are going to conduct a field application experiment using JMAfs 5 research vessels throughout one year to verify the RMNS and clarify the potential problems in the real cruises. To certify the RMNS, the only way now is to conduct laboratory comparison experiments by some laboratories which have higher skill in nutri-ents measurements and have been keeping the better within lab. traceability for long time. between two laboratories. The results of our long-term

OS21B-26 0830h POSTER

Determination of the Organic Complexation of Mercury in Galveston Bay Estuarine Water Using **Competitive Ligand Equilibration** Techniques

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It is well recognized that Hg interacts strongly with organic matter in aquatic systems and that this inter-action is important in its aqueous biogeochemical cyaction is important in its aqueous biogeochemical cy-cling. While Hg-organic ligand interactions have been widely predicted from our knowledge of the environ-mental chemistry of Hg, very little is known about the nature and strength of the interactions from a thermo-dynamic perspective. Competitive ligand equilibration (CLE) approaches have been used to estimate the com-plexation of several metals with natural organic ligands in seawater, but to date, this approach has not heen re- (\dot{CLE}) approaches have been used to estimate the complexation of several metals with natural organic ligands in seawater, but to date, this approach has not been reported for Hg. Here, we report initial results to determine the complexation of Hg by natural organic ligands in estuarine water using CLE approaches. The CLE approach for determining Hg speciation in natural waters is complicated by the fact that Hg-natural organic ligand complex(es) readily extract (> 75%) into non-polar solvents. To apply the CLE approach, a competitive ligand must compete with the natural organic ligand of the non-polar solvent. This approach he competitive metal-ligand complex extracts into the solvent and the natural metal-ligand complex remains in aqueous solution. A ligand which appears to meet these criteria is 1, 2 dimercaptopropanol (BAL). We are currently developing a solvent extraction with CCL4 to isolate the natural organic complexes of Hg from the Hg bound to the competitive ligand. Mercury measurements are conducted using cold vapor atomic fluorescence spectrometry. Mercury natural organic ligand complexes of Hg from the Hg bound to the competitive ligand complex and complex are traction capacity and natural ligand complex that or a fluorescence spectrometry. Mercury natural organic ligand complexation capacity and natural ligand combet to the fluorescence spectrometry of the natural are and the combine dusing conducted using conducted with the matura sample combine do the fluorescence spectrometry of the natural are and the solvent and the hatural organic ligand complexes of the fluorescence spectrometry. Mercury natural organic ligand complexation capacity and natural ligand complexation capacity and natural ligand complex and the solvent are sample combined with the solvent areadined and the tion capacity and natural ligand concentrations are de-termined using linearization techniques applied to Hg titrations of the natural water sample combined with equilibrium modeling of the inorganic solution specia-tion of Hg. Ultra-filtration of natural water samples is being used to investigate the role of Hg-colloid interac-tions in interfering with the CLE and solvent extraction process. Initial results suggest that a major portion of the Hg present in Galveston Bay estuary is complexed by a very small concentration of natural ligand with a conditional stability constant > 10²⁵.

OS21B-27 0830h POSTER

The offset table of IAPSO Standard Seawater and its application to the observational results.

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Tsukuba 305-0052, Japan The comparison of batches of IAPSO standard sea-water (SSW) is important to high quality hydrographic studies. There are a lot of works for batch-to-batch dif-ferences of SSW and the results have been summarized up to P129 as a so-called goffset table". In this study, we examined some of recent batches of SSW and pro-pose the updated goffset tableh up to P140. We apply this table to the observational data such as WOCE ex-pedition data to show how the corrections lead to much better data not only at the regional scale of the Philip-pine Sea but also the basin scale of the Pacific Ocean and Atlantic Ocean. We also apply the table to the most recent result of observations made at the North Pacific (P17 revisit) as a part of Sub-Arctic Gyre Ex-periment granted by Ministry of Education, Culture, Sports, Science and Technology of Japan. To apply the goffset tableh retrospectively means to accept tacily the hypothesis that the batch to batch difference of SSW was not caused only by the aging effect but also, to a considerable extent, the reflection of the gristial offseth. We conducted the comparison of the PSS78 standard KCl solutions made from the different high quality chemicals and found the batch and/ or company dependency. This may be one of the possible cause of ginitial offseth of SSW. The comparison of batches of IAPSO standard sea

OS21B-28 0830h POSTER

Two Analytical Intercomparison Exercises for Iron in Seawater

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United Kingdom Over the past decade, there has been a rapid de-velopment of analytical methods for measuring iron at picomolar levels in seawater. The lack of rigorous inter-comparison exercises and appropriate reference materi-als means oceanographers have little ability to distin-guish between environmental variability and analytical data quality. There is thus an urgent need for stan-dardisation in order to ensure the highest possible in-tegrity, reliability and comparability of reported data. Clearly, with so many sampling and analytical variants in current use, it is difficult to attribute concentration differences to one particular step in the overall process without first determining differences between analytiwithout first determining differences between analytical methods

without first determining differences between analyti-cal methods. This paper presents the results of two analyti-cal intercomparison exercises conducted to determine whether methods were analytically indistinguishable and to attempt to explain any systematic bias in Fe concentrations by examining their operational nature. The first exercise was held at sea during a north-south transect of the Atlantic Ocean (October 2000). To minimise differences from possible sampling contami-nation, one standard underway (1-2 m) protocol was used. Over 200 samples (1 per h) were collected for dissolved Fe (DFe, <0.2 μ m) over 12 d and analysed im-mediately. Four different analytical methods were used, based upon three variants of flow injection with lumi-nol chemiluminescence (FI-CL) and cathodic stripping voltammetry (CSV). During the second exercise, which the Southern Ocean were analysed for DFe (<0.4 μ m) and total dissolvable iron (unfiltered, TDFe), using FI-CL and FI-spectrophotometry (FI-SP). During the shipboard intercomparison (DFe range 0.02-1.61 nM), there was some scatter between sur-face profiles and consistent methodological differences were not always obvious. In general CSV measured

During the shipboard intercomparison (DFe range 0.02-1.61 nM), there was some scatter between surface profiles and consistent methodological differences were not always obvious. In general, CSV measured 0.13 nM more than one FI-CL method, which measured 0.13 nM more than the other two. Statistical analyses (paired two-tailed t-test) show that each analytical method gave significantly different DFe concentrations at the 95% confidence interval. Results from the labbased exercise were more encouraging (Fe range 0.12-0.93 nM). Good agreement was noted for DFe profiles for the stations near the Sub-Antarctic Front and in the Polar Frontal Zone, although there was discrepancy for a station in the Sub-Antarctic Zone. Here, FI-CL recorded 0.07-0.16 nM DFe more than FI-SP at each depth, with greater differences recorded below the mixed layer. TDFe profiles were less consistent, reflecting the added complexity of analysing partially digested particulate material in the samples. FI-SP gave higher TDFe values (~0.12 nM) than FI-CL. Reasons for the differences between analytical methods will be discussed. These include the role of organic complexes and colloids, variability in system sensitivities, nature and pH of sample pretreatment, and ter matrix.

ter matrix.

OS21B-29 0830h POSTER

The Release of Thiols from Marine Algae in Response to Copper Stress the Application of a Solid Phase Extraction and HPLC Determination of Fluorescently Derivatized Thiolic Compounds

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Street, Madison, WI 53706, United States A new method was developed for determination of dissolved thiols in surface using solid-phase extraction to preconcentrate the thiol-monobromobimane (mBBr) derivatives, followed by HPLC analysis of the fluo-rescent tagged compounds. Considerable effort was devoted to optimizing the experimental conditions to maximize derivatization efficiency. The amount of the reducing agent, tri-*n*-butylphosphine (TBP), is criti-cal for the successful total thiol determination. The thiol fluorescence signal could be totally suppressed if the mole ratio between TBP and mBBr is greater than one. We suspect that the tendency of mBBr, with unsaturated-bonds, to attract the electron-enriched TBP could decrease its affinity to thiols. Experiments,

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conducted to determine if the presence of copper affected fluorescence signals of thiols, showed no measurable effect. Good recovery of tagged-thiols spiked in a NaCl (0.5M) solution is obtained by using the Waters HLB reserved phase resin, while blank levels of glutathione, gamma-glutamylcysteine, and phytochelatin-2, are extremely low. For 100 ml samples spiked with 10 nM thiol standards, the average recovery on the 60mg resin cartridge is $42.7\pm4.2\%$, $108.2\pm5.4\%$, $112.7\pm6.5\%$, and $101.5\pm7.8\%$ for cysteine (Cys), glutathione (GSH), gamma-glutamylcysteine(γ EC), and phytochelatin-2 (PC-2). Using this method, a preliminary study showed a clear dose-response release of GSH from a marine algae species, *Thalassiosira weissflogi*, under different concentration levels of copper additions (from 0 to 500 nM). When inoculated in natural seawader different concentration levels of copper additions (from 0 to 500 nM). When inoculated in natural seawa-ter, the release of GSH appears to be inversely related to the concentrations of natural organic matter, reflect-ing the net effects of both the natural organic matter and ligands released from algae on the bioavailability of copper of copper.

OS21B-30 0830h POSTER

RATS: A Robotic Analyzer for the TCO₂ System in Sea Water from Moorings

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RATS is an instrument designed to make the mea RATS is an instrument designed to make the mea-surements needed to determine the speciation of the carbonate system in sea water. Determination of the carbonate system is based on the measurement of TCO₂ and pH. It is designed to operate completely independently and to be deployed for prolonged peri-ods, duration being determined by the current limit of ~1000 sample analyses. RATS can operate at full ocean depth, but testing of components has presently heen limited to the pressure equivalent of 900 m. TCOO be rooted and peth, but testing of components has presently been limited to the pressure equivalent of 900 m. TCO₂ is determined by allowing the CO₂ from an acidified sea water sample to pass through a semi-permeable membrane and react with NaOH. Reaction decreases the NaOH conductivity in proportion to the amount of CO₂ in the sample. pH is measured spectrophotomet-rically after the methods pioneered by Bob Byrne and co-workers. Provision is made for *in situ* standardiza-tion of both TCO₂ and pH at any desired frequency. Presently the instrument is operating in the lab in its integrated form. TCO₂ analyses carried out over a period of ~60 days vary about the mean by slightly less than 0.3%. For shorter time intervals (1-2 days) variation is ~0.2%. The standard deviation of pH mea-surements carried out over a 70+ day period is <.002 pH units. RATS is scheduled for its first ocean tests in Jan-Feb 2002.

. Jan-Feb 2002

OS21C HC: Hall III Tuesday 0830h

Satellite-Measured Ocean Color Variability in the Ocean III

Presiding: A Thomas, University of Maine; C McClain, NASA GSFC

OS21C-31 0830h POSTER

Seasonal and Interannual Phytoplankton Variability in the Gulf of Maine

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Libby Hall, Orono, ME 04469-5741, United States Four years (1997-2001) of ocean color data from the SeaWiFS mission provide the first synoptic quantifica-tion of phytoplankton variability on seasonal and inter-annual time scales for the Gulf of Maine. The clima-tological seasonal cycle shows elevated concentrations (>2.0 mg m⁻³) throughout the year within 30km of the shore and a strong modulation of the amplitude of seasonal cycles over shallow bathymetry in other re-gions. Deeper basins exhibit a canonical Sverdrup sea-sonal cycle with a strong spring bloom (> 2.0 mg m⁻³) gions. Deeper basins existing a calonical Sverdrup sea-sonal cycle with a strong spring bloom (> 2.0 ng m⁻³) in March-April, a fall bloom in October-November and concentrations of ~1.0 in summer and ~0.5 in winter (December-February). Strong tidal mixing over shal-low bathymetry supports elevated concentrations (>

2.0 mg m⁻³) throughout the year on Georges Bank and sustained elevated concentrations throughout the summer over Browns Bank. EOF decomposition of the 4 year time series of monthly composites quantifies the strong dominance of the seasonal cycle (93% of the variance) and shows it to be anomalously weak in late 1997 ance) and shows it to be anomalously weak in late 1997 and throughout 1998. The second mode (2.6%) shows a broad seasonal maximum in late summer (Sept-Oct) strongest in coastal regions and over shallow banks, clearly out of phase with three regions of known strong advection/mixing (the Eastern Maine Coastal Current, the southern Scotia Shelf and Nantucket Shoals). Inter-annual differences in the timing of the spring and fall annual differences in the timing of the spring and fall blooms over much of the domain are consistent with differences in the strength and timing of wind mixing. A 3d coupled biological-physical model captures the dom-inant seasonal variability and spatial pattern over the interior Gulf.

OS21C-32 0830h POSTER

Quantitative, space-based measurements of ocean suspended calcium carbonate with MODIS and SeaWiFS

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Suspended calcium carbonate (particulate inorganic Suspended calcium carbonate (particulate inorganic carbon; PIC) plays a major role in ocean biogeochem-istry and little is known about its oceanic distribu-tion. Recently, algorithms have been implemented with MODIS and SEAWIPS for detection of surface PIC con-centrations. In this paper, we describe the accuracy of the satellite PIC algorithm, based on simultaneous satellite and ship measurements. We also show the first monthly global composites of oceanic PIC, integrated over the euphotic zone (based on satellite estimates of Kpar). Results show relatively high PIC concentrations in specific portions of the North Atlantic and Black Sea, plus in the Antarctic convergence during austral fall. Values of integrated global euphotic PIC concen-trations will be presented.

OS21C-33 0830h POSTER

A multi-year record of bio-optical properties in the Gulf of Maine

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Corvallis, OR 97331-5503, United States Bio-optical data have been compiled from the Gulf of Maine as part of a ship of opportunity program be-tween 1998 and 2001. Data from surface waters were taken aboard the M/S Scotia Prince passenger ferry that crossed between Yarmouth, Nova Scotia, and Port-land, Maine, from May through October of each year. Flexibility in sailing days allowed us unprecedented abilities to sample during clear-sky conditions, when ocean color satellite viewing was optimal. This pre-sentation represents a synthesis of the first three years of measurements, over 3600 sea-truth satellite observa-tions within $\pm 6h$ of a satellite overpass. One clear re-sult is the connection between hydrography and optics; tions within $\pm 6h$ of a satellite overpass. One clear result is the connection between hydrography and optics; there was a prevalence of Case II surface water throughout the entire transect for most of the cruises, which was well correlated to salinity. Even far offshore, using standard criteria for defining particulate and dissolved Case II in character. Also of interest was the significant scattering of the colored dissolved organic matter, which was inversely correlated to salinity in the Eastern Maine Coastal Current. We will present a statistical analysis of volume scattering function shapes

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relative to coastal volume scattering functions of Pet-zold. Errors in SeaWiFS normalized water-leaving radi-ance were highly correlated between wavelengths, with some apparent dependence on the atmospheric models used to derive the radiances. The slopes of the size distribution functions of particulate matter also were compared. There were clear changes in phytoplankton community structure and slope of the particle size dis-tribution, associated with water mass variability.

OS21C-34 0830h POSTER

Physical Forcing of Plumes and Blooms in the Santa Barbara Channel: An Integrated Satellite Approach

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United States Satellite images of sea surface temperature (SST) and ocean color products are used to assess phytoplank-ton dynamics and the extent of terrestrial discharge in the Santa Barbara Channel (SBC). Over three and a half years (October 1997 - June 2001) of AVHRR, SeaWiFS, *in situ* measurements, and supporting data are analyzed. Match-up analyses between *in situ* and satellite data show reliable SST retrievals from AVHRR while improvement is needed for retrievals of SeaW-iFS products in the region. However, spatial/temporal patterns for all SeaWiFS products are qualitatively consistent on both seasonal and episodic time scales. Monthly climatologies of SST, chlorophyll, and water leaving radiance at 555m show blooms associated with high water turbidity and runoff in winter, whereas in late spring blooms are associated with upwelling and cool SSTs. Analysis of monthly composites demon-strates that local winds and terrestrial discharge are well correlated with spatial patterns of upwelling and runoff. A set of empirically defined plume and bloom indices is developed based upon field data sets. These show that plumes are found episodically in response to discharge events and affect 20 - 60% of the SBC whereas blooms occur more regularly and cover 50 - 95% of the fects of upwelling, advection, runoff events, and remote forcing (such as the 1997/1998 El Niño) on blooms and water turbidity will be presented.

OS21C-35 0830h POSTER

Seasonal and inter-annual variability of the particulate backscattering coefficient over the global ocean.

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The global distribution and the annual and inter-annual progression of the particulate backscattering co-efficient, bbp, is analyzed from the Sea-viewing Wide Field-of-view Sensor and inverse modeling. The re-trieved bbp values are compared with in situ measure-ments made in various occanic areas. The seasonal and inter-annual variability of bbp between the years 1998 and 2001 are provided for each major occanic basins. Moreover, the patterns of the remotely sensed bbp are compared to those of the chlorophyll concentration, Chl, in order to determine whether bbp provides addi-tional biogeochemical information. Because bbp is not sensitive to the dissolved material, it can be used as an index of the particulate load that mostly consists of particulate organic carbon, POC, in open occan (bio-genic CaCO3 particles and terrigenous particles may also contribute episodically to bbp). The interpreta-tion of bbp in terms of POC is discussed. The global distribution and the annual and inter-

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