OS62 2002 Ocean Sciences Meeting

COSO2 2002 Ocean Sciences Meet

OS12F-189 1330h POSTER

Growth and Grazing Rates of the Prostomatid Ciliate Tiarina fusus on Red Tide and Toxic Algae

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Kyeong A Seong We investigated growth and grazing rates of the pros-tomatid ciliate Tiarina fusus when feeding on several species of red-tide and/or toxic algae (RTA). T. fusus ingested the dinoflagellates Lingulodinium polyedrum, Scrippsiella trochoidea, Heterocapsa triquetra, Pro-rocentrum minimum, Amphidinium carterae, and the raphidophyte Heterosigma akashiwo, but rarely con-sumed the dinoflagellate Ceratium fusus, and did feed on the dinoflagellate Ceratium fusus, and did feed on the dinoflagellate Prorocentrum micans. T. fusus exhibited positive growth on L. polyedrum, S. tro-choidea, and H. akashiwo. Specific growth rates of L. polyedrum, S. trochoidea, and H. akashiwo before saturating between 500-1,000 ng C ml⁻¹. Maximum specific growth rate of T. fusus feeding on L. polye-drum (0.47 d⁻¹) was much higher than when feeding on S. trochoidea (0.13 d⁻¹) or H. akashiwo (0.10 d⁻¹). Threshold prey concentrations (where net growth = 0) for L. polyedrum, S. trochoidea, and H. akashiwo. were 34-160 ng C ml⁻¹. Maximum ingestion rates of T. fusus on L. polyedrum, S. trochoidea, and H. akashiwo were 23.4, 10.2, and 6.5 ng C predator⁻¹ akashiwo were 23.4, 10.2, and 6.5 ng C $\rm predator^{-1}$ akashiwo were 23.4, 10.2, and 6.5 ng C predator⁻¹ d⁻¹, respectively, while maximum clearance rates were 4.5, 0.2, and 0.6 l predator⁻¹ h⁻¹, respectively. T. fusus exhibited comparable or higher maximum growth, ingestion, and clearance rates than previously reported for the mixotrophic dinoflagellate Fragilidium cf. mexicanum or the heterotrophic dinoflagellates Protoperidinium cf. divergens and P. crassipes, when grown on the same prey species. Grazing coefficients calculated by combining field data on abundances of T. fusus and co-occurring RTA with laboratory data on ingestion rates obtained in the present study suggest that T. fusus sometimes has a considerable grazing impact on the populations of H. akashiwo.

OS12G HC: 319 B Monday 1330h Novel Techniques for Chemical Characterization in Marine Systems I

Presiding: H E Hartnett, Rutgers University; L Minor, Old Dominion University

OS12G-01 1330h INVITED

Comprehensive Two-Dimensional Gas Chromatography: A New Tool for Geochemical Research

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Recent analytical and conceptual advances in chro-matography have led to the development of comprehen-sive two-dimensional gas chromatography (GCxGC), which has a tremendous capability to separate, iden-tify, and quantify trace organic compounds in com-plex geochemical samples. GCxGC uses two different chromatography columns coupled serially by a ther-mal modulator to produce a volatility by polarity sep-aration and distribute compound peaks across a two-dimensional retention time plane. This is in contrast to "heart-cutting" two-dimensional GC, which only al-lows select components in the first column to be re-chromatographed on a second column. The GCxGC two-dimensional separation produces an order of mag-nitude more resolved peaks than traditional GC meth-ods, and the grouping or ordering of the peaks in Recent analytical and conceptual advances in chrochromatographed on a second column. The GCKGC two-dimensional separation produces an order of mag-nitude more resolved peaks than traditional GC meth-ods, and the grouping or ordering of the peaks in the GCxGC chromatogram facilitates the identification of unknown compounds. Furthermore, spatial band compression during modulation has the additional ad-vantage of more than an order of magnitude increase in signal-to-noise ratio, thus improving the detection and quantification of minor components. For exam-ple, we have used GCxGC to investigate the com-position and distribution of alkenones (35 to 39 car-bons) in Black Sea sediments. These compounds dif-fer in chain length, degree of unsaturation, and posi-tion of the ketone group (either methyl or ethyl). In this case, GCxGC analysis resolved and helped ten-tatively identify more of these compounds than was previously thought to exist. New alkenones include C35:2Me, C37:4Et, C37:1Me, C38:4Me, C38:1Et, and C39:4Et. (Alkenone nomenclature is as follows: Cx;y Me or Et, where x is the total number of carbons in the compound, y is the number of double bonds, and Me or Et designates methyl or ethyl ketones, respec-tively). It is especially worth noting that the place-ment of the ketone group by one carbon unit (methyl or ethyl) allowed for discrete bands of these compounds to be resolved. We have also found that GCxGC may be equally adept in determining the alkenone unsatura-tion index in total sediment extracts and hence avoid-ing time-consuming "clean-up" steps when using these compounds for paleothermometry. Another application of GCxGC that we are exploring is resolving the un-resolved complex mixture (UCM) of petroleum hydro-carbons. These compounds are the most abundant, ubiquitous, and understudied class of organic contam-inants in estuarine and coastal sediments. The term UCM is derived from traditional GC and refers to a ubiquitous, and understudied class of organic contam-inants in estuarine and coastal sediments. The term UCM is derived from traditional GC and refers to a hump of unresolved and, hence, unidentified hydrocar-bons in gas chromatograms. Initial efforts have been able to separate and identify discrete bands of normal, branched, one-ring, and two-ring alkanes and one-, two-, and three-ring aromatics. Additional applications fo-cused on natural and anthropogenic organic compounds as well as future efforts will be discussed.

OS12G-02 1345h INVITED

A Molecular Multi-Isotopic Approach for the Study of Organic Matter Cycling in the Oceans

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Over the past several years it has become feasible to measure the isotopic characteristics of individual orto measure the isotopic characteristics of individual or-ganic compounds in complex natural matrices such as marine sediments. In addition to the stable isotopes of carbon, techniques have been developed for exam-ining natural abundance variations in other stable iso-topes $(D/H, \delta^{15}N$ and δ^{18} O), as well as for radiocar-bon at the molecular-level. Here, we illustrate how multi-isotopic measurements on individual compounds can improve our understanding of organic carbon cy-cling in the ocean.

Coupled measurements of δ^{13} C and Δ^{14} C better Coupled measurements of δ^{-1} C and Δ^{-1} C better define organic carbon inputs, both in terms of quantity and mode of transport: δ^{13} C measurements on molecu-lar markers (biomarkers) of marine primary productiv-ity (e.g., algal sterols, fatty acids, alkenones) may help to define environmental conditions and the physiolog-ical status of by the precursor organisms; Δ^{14} C mea-surements on these same computed provide a tempor ical status of by the precursor organisms; Δ^{14} C measurements on these same compounds provide a temporal context for the processes that act upon phytoplanktonic organic matter, from its synthesis in the surface occan to its burial in marine sediments. We have observed, for example, that sediment redistribution processes such as resuspension and lateral advection can significantly "pre-age" organic matter that is preserved in some sedimentary regimes. When paired with similar measurements on vascular land plant biomarkers (e.g., epicuticular waxes, lignin phenols), constraints can be placed on the terrestrial residence time of the latter and on the importance of terrigenous organic carbon in marine sediments, and found this can provide complementary information to that available from carbon isotopes.

In addition to describing this multi-isotopic approach, this paper will describe methods and existing challenges associated with molecular-level radiocarbon measurements, and discuss future analytical developments.

OS12G-03 1400h

A Compound-Specific Isotopic Tracer for Organic Nitrogen Source: Amino Acid 13C Fractionation Patterns

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Most of the organic nitrogen which can be identified at the molecular level in organisms, sinking particles, and dissolved matter is present as amino acids. Un-fortunately, in the geochemical realm, molecular-level amino acid analysis has had limited ability to provide source information because overall amino acid compo-sitions from different sources are too similar. On the isotopic level, however, distinct biochemical pathways associated with differing metabolic processes result in characteristic stable isotopic fractionation patterns. In

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proteinaceous materials such shifts are large, in the range of 5-20 among individual amino acids. Photosyn-thetic metabolism results in a highly reproducible pat-tern, while in comparison material from heterotrophic sources produce distinct and repeatable offsets in sev-eral amino acids. Thus the overall distribution of amino reid for the patient particip metabolism. eral amino acids. Thus the overall distribution of amino acid fractionation in the main protein amino acids of-fers a powerful potential tool to discriminate metabolic and organism-level source of organic nitrogen. We ex-amine this technique using oceanic samples of dissolved and particulate material from the central Pacific, focus-ing on the potential to distinguish nitrogenous materi-als from prodervotic vs. aud autotrophic als from prokaryotic vs. eukaryotic, and autotrophic vs. heterotrophic sources.

OS12G-04 1415h

Combining Molecular Level Anlayses and Nuclear Magnetic Resonance Spectroscopy (NMR) to Characterize Dissolved Organic Nitrogen (DON) in the Marine Environment

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The use of ultrafiltration to isolate milligram quan The use of ultrafiltration to isolate milligram quan-tities of marine dissolved organic matter (DOM) has greatly advanced our understanding of the chemical components dominating this reservoir. We are now able to combine sophisticated analytical techniques such as NMR, and mass spectrometry (MS), with more tradi-tional molecular level analyses, to elucidate the chemi-cal structure and, as a consequence, the cycling of dif-ferent components of DOM in the ocean. Here we de-oribe the use of them taphongue to identify the abami. ferent components of DOM in the ocean. Here we describe the use of these techniques to identify the chemical species that dominate the dissolved organic nitrogen (DON) reservoir in both the surface and deep ocean. Dissolved organic nitrogen is the largest reservoir of fixed nitrogen in the surface ocean. Based on the depth profiles of DON concentration in the open ocean, it appears that approximately half of this reservoir is reactive on relatively short timescales. Using a combination of ¹H, ¹³C and ¹⁵N NMR, coupled with selective chemical degradation techniques and molecular level analyses of monosaccharides, amino acids and acetate, analyses of monosaccharides, amino acids and acetate analyses of monosaccharides, amino acids and acetate, we are able to show that the reactive high molecular weight DON is present within surface derived polysac-charides that are up to 50 % N-acetyl amino sugars. The non-reactive DON appears to be a complex mix-ture of compounds dominated by proteins that are re-sistant to chemical hydrolysis. We are also in the process of isolating and further characterizing the individual components of the DON reservoir using a combination of biochemical techniques such as electrophoresis and colection computer the derenderity of the second secon selective enzymatic degradation.

OS12G-05 1430h

Phosphorus Cycling and Regeneration in the Ocean An Evaluation Using 31P-NMR

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University, Stanford, CA 94305-2115, United States Phosphorus (P) is an essential nutrient utilized by all living organisms. It has been recognized as a lim-iting nutrient in some oceanic systems and is possi-bly the ultimate limiting macronutrient for marine pro-ductivity over long time scales. Phosphate availabil-ity to phytoplankton growth, and thus the effective-ness of the biological pump depends on input of P to the euphotic zone via upwelling of deeper water and from river sources in coastal environments. Despite the recognition of the important role P plays in controlling marine productivity, relatively little is known about P cycling and regeneration within the ocean (Benitez-Nelson, 2000). Basic questions still remain unanswered such as: What phosphatic compounds are preferen-tially regenerated? How does P regeneration vary spa-tially and temporally? and, What determines the depth tially and temporally? and, What determines the depth

and rate of P regeneration? Identification of the P compounds in the ocean is necessary to determine the origin, transformation, and potential bioavailability P, and to understand the structure of P compounds that ultimately are incorporated in the geological record. We have assessed changes in the molecular compo-

We have assessed changes in the molecular compo-sition of the particulate phosphorus pool in various wa-ter depths and oceanic settings using 31P-NMR spec-troscopy. Particulate matter was analyzed from sedi-ment traps from 3 different oceanic regimes: a coastal environment, an open ocean environment, and a polar environment. These different oceanic regimes represent areas spanning a wide range of productivity, soluble re-active phosphate concentrations in the euphotic zone, eas-surface temperatures and planktonic communities sea-surface temperatures, and planktonic communities sea-surface temperatures, and planktonic communities. Preliminary results indicate that the oceanic particu-late P pool is composed of: orthophosphate, phospho-monoesters, phosphodiesters, pyrophosphate and phos-phonates. Variability in the relative contribution of these components to the total P pool has been observed both spatially and with depth/age. Results will be interpreted in terms of preferential formation and/or degradation of specific organic compounds. The in-formation gained with respect to the distribution and change of bond structure in these samples is valuable for assessing the potential reactivity of different P com-pounds. for asse pounds

OS12G-06 1445h

Characterization and Preliminary Field Trials of a Bioluminescent Reporter of Iron Bioavailability

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The role of Fe in the control of productivity and community structure of planktonic assemblages is well established in oceanic HNLC areas and several large lakes. Dissolved iron concentrations can be very low in the surface layer and dependent on both the pres-ence of dissolved organic ligands and light penetra-tion. The limitation of heterotrophic bacterial pro-duction by Fe availability in these systems is still de-bated due to conflicting field results showing either a primarily organic carbon stress or Fe stress. There-fore, the development of an approach to distinguish between organic carbon and iron bioavailability repre-sents one of the significant challenges for the Fe field research. We have developed a biomolecular tool to track the response of heterotrophic bacteria to ambiresearch. We have developed a biomolecular tool to track the response of heterotrophic bacteria to ambitrack the response of heterotrophic bacteria to ambi-ent Fe concentrations. Bioluminescent reporters, con-structed by fusing the [pa/se promoter of Escherichia coli (a siderophore biosynthesis gene) to a luxCDABE transposon, were integrated into the chromosome of a halotolerant Pseudomonas putida as well as marine het-erotrophic prokaryotes. Light production from the lux cassette is thus linked to the activation of the report bacteriums high-affinity. Fe transport systems (an indi-cation of Fe-limitation), making it a useful indicator of Fe availability. The P. putida bioreporter has been suc-cessfully tested both in lab and field studies. Labora-tory cultures were grown under degrees of Fe limitation generated by the addition of concentrations (0 to 150 nM) of well-characterized chelators (desferrioxamine B [DFB], ferrichrome) as well enrichment with inorganic Fe (5 to 20 nM). Preliminary characterizations suggest nM) of well-characterized chelators (desferrioxamine B [DFB], ferrichrome) as well enrichment with inorganic Fe (5 to 20 nM). Preliminary characterizations suggest the growth rate of the bioreporter decreased significantly by a factor 2 and 3 for DFB and ferrichrome additions relative to iron-replete treatments. However, a relative light increase of 25% and 40% was observed for the same chelators between iron-limited and iron-rich media. These results confirm the hypothesis that these bioreporters can be used to monitor Fe availability. Field studies in a low Fe system (the central basin of Lake Erie) were performed with DFB added to trace metal clean collections of water at concentrations of 0, 12, 20, 30, 40 and 50 nM. Bioreporters expressed Fe-stress (aa 200% light increase) at concentrations of DFB > 30 nM, (but not < 20 nM DFB) indicative of the removal of bioavailable Fe, with increasing light produced at both 40 and 50 nM DFB. In a similar experiment with 0.2 - μ m prefiltered water a dramatic increase in light production (aa 600% relative to 0 nM DFB controls) was observed at the lowest (12 nM) concentration of chelators, more than 2-fold higher than the unfiltered water sample. These results suggest that removal of the particulate size class forced cells into Fe-stress more rapidly when DFB was added, suggesting that a significant component of the bioavailable Fe flux is particulate in nature.

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OS12G-07 1520h INVITED

Identification of Metal-Complexing Ligands in Sea Water using Electrospray Mass Spectrometry

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Section, 9860 West Saanich Road, Sidney, BC V8L 4B2, Canada Organic complexation affects the bioavailability and distribution of trace metals in seawater. Electrochemi-cal methods have shown that Cu^{2+} forms strong com-plexes with organic ligands in the surface ocean, and several laboratory studies have shown that copper-specific ligands are released by photosynthetic ma-rine organisms in response to elevated copper concen-trations. These compounds need to be isolated and identified before their role in the biogeochemical cy-cling of trace metals can be fully understood. Elec-trospray mass spectrometry (ESI-MS) provides infor-mation about the structure of dissolved organic lig-ands and their metal-ion complexes that cannot be obtained using electrochemical methods. Using 8-hydroxyquinoline and tannins as model ligands, we'll show that ligands can be characterized by ESI-MS and that the location of the copper binding site can be de-termined in complex molecules. We'll also present re-sults on copper binding ligands from coastal waters of British Columbia, extracted using immobilized Cu^{2+} . ion affinity chromatography (IMAC), monitoring the UV absorbance at 255 nm during elution. Salts and residual metal ions were removed on line using a cus-tom made XAD-16 column. Ligand concentrations were higher at low salinity and in surface waters, suggest-ing that these ligands are produced in surface waters and/or delivered to the region via the Fraser River. Analysis of the extract with highest UV absorbance identified Cu^{2+} ligand of molecular weight 259. The mass and isotopic distributions are consistent with a dipeptide containing two metal-binding amino groups. mass and isotopic distributions are consistent with a dipeptide containing two metal-binding amino groups. This result is consistent with the findings of other studies attempting to characterize Cu^{2+} ligands in seawater. The structure of the identified ligand is similar to that of rhodotorulic acid (a microbial siderophore), indicating that small peptides and related compounds can act as strong, specific metal chelators in natural waters.

OS12G-08 1535h INVITED

Characterization of Thiols in Seawater by Electrospray Mass Spectrometry of Monobromobimane Derivatives

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Recent studies suggest that certain thiols, such as glutathione, are present in appreciable quantities in surface waters where they may be important metal chelators. Much of this data was obtained by cathodic stripping voltammetry. Here we show that while this is a useful method that can provide additional information about metal complexation, it is not selective for specific thiols, and indeed the species detected in general do not behave like glutathione, the compound frequently inferred to be the major thiol present. HPLC of derivatized thiols is a more powerful approach, provided that standards are available for all the thiols in the sample, to enable identification based on retention times alone. However, some peaks in real samples do not correspond to common standard compounds. Here we report the use of electrospray mass spectrometry coupled with HPLC for thiols derivatized with monobromobimane, and we show how this approach provides useful structural information for a variety of thiols found in natural waters. Recent studies suggest that certain thiols, such as

riety of thiols found in natural waters.

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OS12G-09 1550h

Chemical Characterization and **Bioavailability of Dissolved Organic** Matter using Atmospheric Pressure Electrospray Ionization Mass Spectrometry

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Dissolved organic matter (DOM) is a complex mix-ture of organic compounds and is an important source of C and N to rivers and estuaries. As such, DOM can fuel bacterial metabolism, but the reactivity of DOM varies by source and season. Until recently, our un-derstanding of DOM reactivity and bioavailability has been limited to bulk-level analyses or to some specific fraction of the DOM pool. We present new results us-ing Atmospheric Pressure Electrospray Ionization Mass spectrometry to characterize the bioavailability of DOM from various sources. This analytical technique pro-vides information on molecular weight, abundance and acid/base properties. We demonstrate differences in the reactivity of individual compounds and chemically characterize the bioavailabile and refractory pools of DOM before and after microbial degradation. Samples from a variety of aquatic sources (forested, agricultural and urban) had different molecular weight distributions or 'fingerprints', as well as different degrees of bioavail-ability. In an experiment with urban stormwater, ap-proximately 40% of the DOM compounds were bioavail-able, and almost all of the available compounds ex-hibited similar chemical characteristics. This approach has extended our understanding of DOM processing in aquatic systems by providing molecular-level informa-tion on the previously uncharacterized complex mixture of DOM compounds. Dissolved organic matter (DOM) is a complex mixof DOM compounds.

OS12G-10 1605h

Techniques for the Analysis of Intact Protein in Marine Sediments

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Proteins represent the largest biochemical class of compounds identified in marine sediments. Traditional thought suggested that proteins were very labile, and therefore unable to withstand diagenetic transport and alterations. The majority of studies to date on sedi-mentary protein concentrations and dynamics have re-lied on the analysis of protein subunits, i.e. amino acids or peptide bonds. This has limited, or eliminated, all information on the size, type and source of the pro-tein or peptide along with any sequence information. The primary focus of our research has been to develop techniques to look at intact proteins and peptides in sweater, sediments, and pore water. Recent advances in proteometrics now allow for a higher tolerance for salts and impurities from biological matrices and also permit greater sensitivity for protein analysis. With the devel-opment and use of LC-ESI-MS-MS and MALDI-TOF-MS systems we are able to observe cleavage patterns of known proteins My natural bacterial assemblages. Most compounds identified in marine sediments. Traditional Mb systems we are able to observe cleavage patterns of known proteins by natural bacterial assemblages. Most recently, we have also extracted and analyzed naturally occurring proteins from sediments and seawater from the Washington coast. Results to date indicate a pre-dominance of extractable protein in the mid-size range of 30kDa.

OS12G-11 1620h

The Application of Electrospray Ionization FT-ICR Mass Spectrometry to the Study of Natural **Organic Matter**

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Molecular level characterization of natural organic matter has been elusive due to the inherent complexity of natural organic mixtures and the fact that individ-ual components are polar and macromolecular. Elec-trospray ionization (ESI) is a "soft" ionization tech-nique that ionizes polar compounds from aqueous solu-tions prior to acceleration into a mass spectrometer. In this study, we have combined ESI with an ultra-high-resolution mass spectrometer, the Fourier transform ion cyclotron resonance mass spectrometer (FT-ICR MS), to examine individual molecules within a variety of nat-ural organic mixtures. With the high resolution of the FT-ICR MS (<1 ppm error in mass accuracy), we have been able to resolve >9,000 compounds within the 300-1000 Da mass range. At low m/z, the resolution is high enough to assign exact molecular formulas allowing spe-cific components of these mixtures to be identified. In addition to molecular identification, we can now use ESI FT-ICR MS to examine molecular-level changes in different organic mixtures as a function of relevant geo-processes, such as microbial alterations and photo-chemistry. In this presentation, we will highlight the capabilities of the instrument by showing molecular-level resolution mass spectra of different organic mix-tures such as humic substances, riverine DOM and bac-terial exudates. In addition, we will present a exam-ple of the potential of this technique for molecular-level comparisons as a function of photochemical alteration of riverine DOM. Molecular level characterization of natural organic

OS12G-12 1635h

Development of a High Specificity ELISA Assay to Trace the Source and Fate of Biopolymers in the DOC Pool.

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Dissolved organic carbon (DOC) is the largest pool of active organic carbon on earth. Understanding the cycling of DOC is crucial to predict ocean processes ranging from food web dynamics to climate change. To cycling of DOC is crucial to predict ocean processes ranging from food web dynamics to climate change. To investigate the complex degradation dynamics taking place in the water column it is critical to develop probes of high sensitivity and specificity that can identify the source and trace the fate of specific moieties found in DOC. The unicellular alga Phaeocystis (Prymnesio-phyte) is a leading contributor of biopolymers to the dissolved organic carbon pool (DOC). During spring, subpolar and polar oceans blooms of Phaeocystis re-lease huge masses of polymer gels. Most of this material is partially processed by bacteria and other microbial species. However, the final fait of the bulk of Phaeo-cystis primary production remains uncertain, and the contribution of Phaeocystis polymeric material to the global DOM pool is unknown. We developed an ELISA assay (enzyme linked immunosorbent assay) to quantify the concentration and contribution of Phaeocystis ex-tracellular polymers to the global DOC pool. The con-centration of Phaeocystis polymer found in field sam-ples of DOM taken from the Ross Sea, the North Water Polynia and the Gulf of Alaska can range from 0.33 uM to 20 uM carbon. Our results indicate that ELISA pro-vides a powerful highly specific method with nanomolar sensitivity to identify and measure the complex moi-eties found in DOM. (Supported by grants from the US DOE-BIOMP Program and NSF Division of Bioengi-neering Biocomplexity Program) OS12H HC: 314 Monday 1330h

Interactions Between Macro- and Microorganisms in Aquatic Sediments

Presiding: R Haese, Utrecht University; J Kostka, Florida State University

OS12H-01 1330h

Interactions Between the Cold Seep Tubeworm, Lamellibrachia cf luymesi, and interstitial Sulfate Reducing Bacteria: A Perpetual Motion Machine?

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Lamellibrachia cf luymesi is similar to its hy-drothermal vent relatives in that it has no mouth, gut or anus and obtains its nutrition from symbiotic chemoautotrophic sulfide-oxidizing bacteria. Also like gut or anus and obtains its nutrition from symbiotic chemoautotrophic sulfide-oxidizing bacteria. Also like its vent relatives, L. luymesi supplies sulfide to its sym-bionts using special hemoglobins in its vascular and ocelomic fluids that have a very high affinity and ca-pacity to bind and carry sulfide. However, unlike the well-studied East Pacific Rise hydrothermal-vent tube-worms, adult L. luymesi live in an environment where sulfide is normally not detectable around its gas ex-change organ, the obturacular plume. We have recently demonstrated that under some conditions a posterior of taking up sulfide at rates sufficient to fuel net in-organic carbon uptake, or in other words, auctorophy. We have also found that very high (millimolar) levels of sulfide are often present deep in the sediment, around the buried posterior ends of the worms. Where does this sulfide come from and how is it re-Plenished? It is unlikely to come form deep sources as sulfide. Furthermore, these pools are present agreater depths than can be explained by diffusion and reduc-tion of seawater sulfate. We will present a hypothesis or a mechanism to deliver the high sulfide flux necces-sary to supply the roots of the tubeworm aggregations in the Gulf of Mexico, current evidence supporting this hypothesis, and the results of models that constrain the fluf of sulfice required by the aggregations and the volume of sediment that supplies this sulfide.

OS12H-02 1345h

Comparison of Clambed and Microbial Mat Habitats at Eel River Methane Seeps, Northern California Margin: Microbiological, Geochemical and **Biological interactions**

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³ UFZ Center for Environmental Research Leipzig Halle, Permoserstrasse 15, Leipzig 014318, Germany During 2 cruises (2000 and 2001) we investigated the methane seeps in the Eel River Basin on the north-ern California margin (500 m). The study area was characterized by 2-m to 10-m patches of bacterial mats and clambeds occurring in close proximity to each other. Our combined studies focused on the microbial processes and resulting geochemical gradients and their role in structuring the different benthic assemblages. We measured concentration profiles using microelec-trodes (oxygen, sulfide), applied conventional chemi-cal methods (sulfate, methane), and carried out in-situ

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